

## ELECTROCHEMICAL SYNTHESIS OF INORGANIC COMPOUNDS%0A

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 Electrochemical synthesis (inorganic synthesis) offers several advantages over conventional synthesis and often provides the only viable route. Electrosynthetic processes can often be run under milder operating conditions with fewer chemical reagents.

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Electrochemical synthesis of inorganic compounds is a relatively unknown field. The successful, large industrial processes, such as chlorine-caustic production, are well known, but the large number of other compounds that have been synthesized electrochemically are much less appreciated, even by electrochemists and inorganic chemists.

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The organonickel complexes are organometallic compounds containing a Ni-C-bond ( $\pi$ -complexes). These species are very reactive and have been mainly characterized as the intermediates of catalytic processes of cross coupling and homocoupling involving organic and elementoorganic substrates such as organic halides, chlorophosphines

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**Direct Electrochemical Synthesis of Inorganic and ...**

The method of direct electrochemical synthesis consists of oxidizing a metal anode in a non-aqueous solution containing a ligand (or ligand precursor) to produce the appropriate inorganic or organometallic compound. In many cases, the product precipitates directly in the cell, making for easy

**Electrochemical routes for industrial synthesis - SciELO**

Inorganic synthesis in ionic liquids . Ionic liquids are not new; some of them have been known for many years, for instance [EtNH<sub>3</sub>]<sup>+</sup>[NO<sub>3</sub>]<sup>-</sup>, which has a melting point of 12 °C, was first described in 1914.<sup>99</sup> The early history of ionic liquid research was dominated by their application as electrochemical solvents. One of the first recognised uses of ionic liquids was as a solvent system for

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Description : Electrochemical synthesis of inorganic

compounds is a relatively unknown field. The successful, large industrial processes, such as chlorine-caustic production, are well known, but the large number of other compounds that have been synthesized electrochemically are much less appreciated, even by electrochemists and inorganic chemists. The last comprehensive book on this subject